EVALUATION OF ROOT-MEAN-SQUARE RADIUS OF GYRATION AS A PARAMETER FOR UNIVERSAL CALIBRATION OF POLYSACCHARIDES

MARSHALL L FISHMAN, WILLIAM C DAMERT, JOHN G PHILLIPS, AND ROBERT A BARFORD US Department of Agriculture, 600 East Mermaid Lane, Philadelphia, PA 19118 (USA) (Received April 17th, 1986, accepted for publication, September 30th, 1986)

ABSTRACT

High-performance, size-exclusion chromatography columns were calibrated in average root-mean-square radii of gyration (\overline{R}_{gz}) by a combination of commercial "narrow" pullulan and "broad" dextran standards. The nonlinear calibration-curves were fitted by a computer-aided, iterative, least-squares procedure. Values of \overline{R}_{gz} , obtained from a point-by-point transformation of the respective pullulan and dextran chromatograms by utilizing universal calibration, were compared with inputted \overline{R}_{gz} calibration values. For standards ranging in \overline{R}_{gz} value from 20.1 to 389 Å, the accuracy ranged from 1 to 15.3%. Furthermore, from relationships in the literature, \overline{R}_{gz} values were transformed to \overline{M}_{w} . These values of \overline{M}_{w} were comparable to, but generally less accurate than, \overline{M}_{w} values from direct, molecular-weight calibration.

INTRODUCTION

In the course of characterizing citrus pectins, it became obvious that existing methods of high-performance size-exclusion chromatography (h.p.s.e.c.) are unsatisfactory, because pectin undergoes concentration-dependent disaggregation $^{1-3}$. Direct molecular-weight characterization of disaggregating macromolecules is often not feasible, because their molecular weight is concentration-dependent, and absolute techniques required for obtaining their molecular weights operate at higher concentration than does h.p.s.e.c. Conventional characterization by universal calibration is often impossible, because intrinsic viscosities cannot be obtained in cases where nonlinear extrapolations to infinite dilution are required. The root-mean-square radius of gyration (\bar{R}_{gz}) is a better parameter for calibration, because a linear extrapolation of \bar{R}_{gz} against an unknown concentration is unnecessary.

For a series of pullulans and dextrans having narrow molecular-weight distributions, Kato et al.⁴ demonstrated the validity of the universal-calibration principle in aqueous h.p.s.e.c Namely, they found that, by plotting $\log \overline{R}_{gz}$ against retention time, the data for pullulans and dextrans fall on the same straight line. We now demonstrate the validity of universal calibration with a series of commer-

cially available, well characterized dextran standards having broad molecular-weight distribution and of pullulan standards having narrow molecular-weight distributions. Moreover, molecular weights obtained from empirical relationships between $\bar{R}_{\rm gz}$ and molecular weight are compared with molecular weights from direct calibration

EXPERIMENTAL

Materials — Dextrans were purchased from Pharmacia Chemical Co., Piscataway, NJ, and pullulans, from Polymer Laboratories, Inc., Amherst, MA. Table I contains molecular weight data obtained from these suppliers

Chromatographic analysis. — Chromatography, sample preparation, and determination of column void and total volume have been described elsewhere⁵. H.p.s.e.c. was performed either in a column (30 × 0.39 cm) of Waters E-1000 or E-linear micro-Bondagel. Of a sample (0.3 mg/mL) there was injected 20 μ L. The mobile phase, either 0.05m or 0.1m NaCl, was stirred magnetically in the reservoir, and the column was wrapped with soft-foam insulator. The chromatograph was housed in a temperature-controlled room at 23 ±1°. Flow rates were measured by a bubble injected into a calibrated measuring-pipet connected to the exit line of the chromatograph⁶. The pump was set at a nominal flow-rate of 0.5 mL/min. Long-term flow-rates were measured to within ±2% of the nominal value. Over any short period, flow rates were precise to ±0.3%. Generally, 3 consecutive peak-maxima agreed to within 2 s

Peaks emerging from the s.e. chromatograph were detected by differential refractive index ($\Delta r.i.$). Analog signals were digitized at the rate of 150 points per min, in a remote location, by a Modcomp 7861 computer equipped with an analoginput subsystem. Chromatograms could be displayed on the cathode-ray tube of an

TABLE I

MOLECULAR WEIGHT STANDARDS

Dextrans					Pullulans		
Sample	$\overline{\mathbf{M}}_{w}/\overline{\mathbf{M}}_{n}^{a}$	$\overline{\mathrm{M}}_{\mathrm{w}} \times 10^{-3}$	$M_p \times 10^{-3b}$	WT% $\leq \overline{M}_{w}^{c}$	Sample	$\overline{\mathbf{M}}_{n}/\overline{\mathbf{M}}_{n}$	$\overline{\mathrm{M}}_{\mathrm{w}} \times 10^{-3}$
T-10	1 63	9 3	7 0	49 0	P-5	1 07	5 8
T-20	1 50	22 3	16 3	70 6	P-10	1 06	12 2
T-40	1 54	44 4	30.0	71 0	P-20	1 07	23 7
T-70	1 65	7 0 0	40 0	57 0	P-40	1 09	48
T-110	1 39	106	69 0	52 0	P-100	1 10	100
T-250	2 25	253	75 0	69 0	P-200	1 13	186
T-500	2 91	532	140	70 8	P-400	1 12	380
T-2000		2000		_	P-800	1 14	853

 $^{{}^}a\overline{M}_{\mathbf{w}}$ is weight average molecular weight, $\overline{M}_{\mathbf{n}}$ is number average molecular weight ${}^bM_{\mathbf{p}}$ is molecular weight at peak maximum ${}^cWT\% \leqslant \overline{M}_{\mathbf{w}}$ is weight fraction with molecular weight equal to or less than, the weight-average molecular weight

TABLE II	
CALIBRATION	-CURVE CONSTANTS

Variables				Consta	ints ^a				
Col b	Conc.c	St d	Sp «	K ₁	K ₂	b _o	b _i	b ₂	b ₃
E l.m.b	0 05	P,D	$\bar{R}_{\sigma \sigma}$	0 205	0 728	3 89	-9 055	12 95	-7 35
E-1000	0 05	P,D	$\bar{R}_{\bullet \tau}^{\omega}$	0 337	0.815	4 04	-8 167	12 02	-6 89
Elmb	0 10	P,D	$\bar{R}_{\sigma \sigma}^{\sigma \sigma}$	0 258	0 760	4 93	-13 67	20 06	-109
E-1000	0.10	P,D	\bar{R}_{σ}	0 356	0 811	4 205	-8.332	11.71	-654
Elm.b.	0.05	P	$egin{array}{l} ar{R}_{gz} \ ar{R}_{gz} \ ar{R}_{gz} \ ar{R}_{gz} \ ar{M}_{w} \end{array}$	0 211	0 607	8 568	-18 94	30 92	-19.4
Elmb.	0 05	D	$\overline{M}_{\mathbf{w}}^{"}$	0 256	0 675	7 660	-8 411	2 135	3 315
E-1000	0 05	P	$\overline{M}_{\mathbf{w}}^{"}$	0 259	0 912	8 075	-12 14	17 89	-10 5
E-1000	0 05	D	$\overline{M}_{\mathbf{w}}^{"}$	0 453	0 815	6 749	-3 197	1 382	-1 25
Elmb	0 10	P	$\overline{M}_{\mathbf{w}}^{"}$	0 274	0 647	11 42	-3363	56 60	-341
Elmb	0 10	D	\overline{M}_{w}	0 352	0 556	1 156	39 72	$-108\ 3$	84 82
E-1000	0 10	P	$\overline{M}_{\mathbf{w}}^{"}$	0 356	0 872	8 314	-11 92	15 65	-8 55
E-1000	0 10	D	$\overline{M}_{\mathbf{w}}^{"}$	0 497	0 811	6 495	-1 112	-2 037	0 269

"Symbols defined by Eqs 1-3 bE l m b. = E-linear micro-bondagel column; E-1000 = E-1000 miro-Bondagel column "Molar concentration of sodium chloride in mobile phase "Standards P = pullulan, d = dextran "Size parameters

Admiral Model 5 Dumb Terminal immediately after the run. User-interactive processing of data, including specification of peak base-line, maxima, and integration limits by cursor, was accomplished with software that was developed in-house.

Averages of \overline{R}_{gz} or molecular weight were calculated by transforming partition coefficients (K_{av}) , point by point, to either \overline{R}_{gz} or molecular-weight values, and summing the appropriate integrals. Integrations were made by use of a trapezoidal algorithm. Transformations were obtained from the following calibration curves.

$$ln(Y) = a_0 + a_1 K_{av}, \text{ when } K_1 > K_{av}$$
 (1)

$$\ln(Y) = b_0 + b_1 K_{av} + b_2 K_{av}^2 + b_3 K_{av}^3, \text{ when } K_1 < K_{av} < K_2$$
 (2)

$$ln(Y) = c_0 + c_1 K_{av}, \text{ when } K_{av} > K_2,$$
 (3)

where, Y is \overline{R}_{gz} or \overline{M}_{w} .

With the aid of Eqs. I-3, the best regression-line is fitted, using values of Y and $K_{\rm av}$. The constants a_0 , a_1 , c_1 , and c_0 are constrained to make the calibration curve and its first derivative continuous at K_1 and K_2 , which are points of intersection between Eqs. I, 2 and 2, 3, respectively, were chosen to minimize the sum of the residuals squared. The constants b_0 , b_1 , b_2 , and b_3 , governing that portion of the calibration curve with $K_1 < K_{\rm av} < K_2$, were obtained by nonlinear regression, using the Gauss-Newton algorithm. The calibration curves were cubic polynomials with straight lines at the ends. The values of K_1 , K_2 , a_0 , a_1 , a_2 , and a_3 for the columns used in this study are listed in Table II.

For "narrow" pullulan molecular-weight standards, values of partition

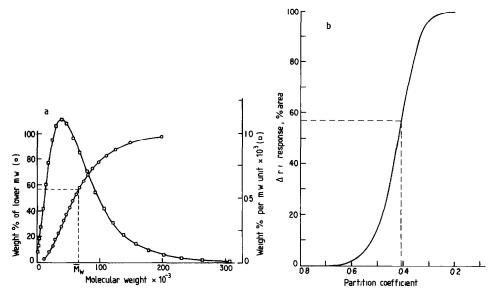


Fig 1 (a) Molecular-weight distribution of dextran T-70 (b) Refractive index integral distribution curve for T-70 dextran from h p s e c on E-linear micro-Bondagel column, with 0 1M NaCl as mobile phase

coefficient corresponding to the peak maximum of the differential, refractive-index trace ($\Delta r.i.$) of the chromatogram were correlated with the z-average radii of gyration (\overline{R}_{gz}) obtained from the literature⁴. Integral distribution curves supplied by the manufacturer gave weight percentage values corresponding to \overline{M}_{w} , the weight-average molecular weight for each broad dextran standard; as shown by Fig. 1a, the value is 57% for T-70 dextran. For each dextran standard, the weight percentages in Table I were equated with area percentages from the $\Delta r.i.$ trace of the corresponding dextran chromatogram, so that K_{av} on the chromatograms could be correlated with \overline{M}_{w} . A typical, integral-distribution chromatogram for T-70 dextran is shown in Fig. 1b.

THEORY

Eqs. 4-7 are four variations of the universal-calibration principle9.

$$\log f(V_h) = f(r.t.) + constant$$
 (4)

$$\log\left[\eta\right]\overline{M}_{\mathbf{w}} = aV_{\mathbf{r}} + b \tag{5}$$

$$\log\left[\eta\right]\overline{M}_{n} = aV_{r} + b \tag{6}$$

$$\log \bar{R}_{\rm gz} = aK_{\rm av} + b \tag{7}$$

Eq. 4, the most general statement of the universal calibration principle (u.c.p.) equates some function of the logarithm of the hydrodynamic volume, (V_h) , with a function of the column retention-time (r.t.). Eqs. 5 and 6, the most common state-

ments of the u.c.p., relate the product of the intrinsic viscosity $[\eta]$ and the weight (\overline{M}_w) or number average (\overline{M}_n) molecular weight to the retention volume (V_r) . Both of these statements require either viscosities obtained by extrapolation to zero concentration for the unknown, or Mark-Houwink constants⁷. Eq. 7, invoking u.c. p, enables the \overline{R}_{gz} of the unknown to be obtained from the \overline{R}_{gz} of known standards. If a relationship between \overline{R}_{gz} and molecular weight is available for the unknown, the molecular weight can be obtained. Moreover, in the case of rod-like molecules, the molecular weight of the unknown can be obtained directly from \overline{R}_{gz} measurements, provided that the virtual bond-length of the monomer unit and the monomer-residue weight are known^{3,8}.

In the case of pullulan, Eq. 8 relates $\overline{M}_{\rm w}$ to the z-average radius of gyration, $\overline{R}_{\rm gz}$ (ref. 4), whereas, in the case of dextran, Eqs. 9 and 10 relate⁷ $\overline{M}_{\rm w}$ to $\overline{R}_{\rm gz}$ Pullulan

$$\overline{M}_{w} = 37.4 R_{gz}^{1.68} \tag{8}$$

Dextran

$$R_{\rm gz} \le \frac{103 \text{ Å}}{\overline{M}_{\rm w}} = 11.9 R_{\rm gz}^2$$
 (9)

$$R_{\rm gz} > \frac{103 \text{ Å}}{M_{\rm w}} = 2.62 R_{\rm gz}^{2.32}$$
 (10)

The larger exponents for R_g in Eqs. 9 and 10, as compared to 8, indicate that, for isomolecular weights, dextrans are more compact than pullulans.

RESULTS AND DISCUSSION

In Fig. 2a are typical, overlayed chromatograms from separate runs of a series of pullulans chromatographed on a column of micro-Bondagel, E-1000, whereas, in Fig. 2b are comparable data for a series of dextrans. For both sets of chromatograms, the mobile phase was 0.05M NaCl. Comparison of Figs. 2a and 2b confirmed that the dextrans are more polydisperse than the pullulans, and that E-1000 columns discriminate better between the high-molecular-weight standards than the low ones

In Fig. 2c are typical, overlayed chromatograms from separate runs of a series of pullulans chromatographed on a column of E-linear micro-Bondagel, whereas, in Fig. 2d are comparable data for a series of dextrans. For both sets of chromatograms, the mobile phase was 0 1m NaCl. Comparison of these two Figures also confirmed the greater polydispersity of dextrans over pullulans. Furthermore, unlike E-1000 columns, E-linear columns discriminate between the lower-molecular-weight standards better than between the higher ones. Fig. 3a and 3b demonstrate

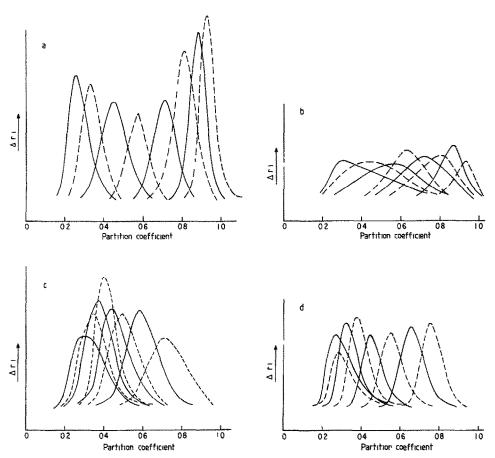


Fig. 2. (a) Overlayed chromatograms of pullulan standards P-800, P-400, P-200, P-100, P-40, P-20, P-10, and P-5. [Dotted and solid lines delineate adjacent chromatograms Column, E-1000 micro-Bondagel; mobile phase, 0.05M NaCl, detector, refractive index] (b) Overlayed chromatograms of dextran standards, T-2000, T-500, T-250, T-100, T-70, T-40, T-20, and T-10. [Dotted and solid lines delineate adjacent chromatograms. Column, E-1000 micro-Bondagel, mobile phase, 0.05M NaCl; detector, refractive index.] (c) Overlayed chromatograms of pullulan standards P-800, P-400, P-200, P-100, P-40, P-20, P-10, and P-5. [Dotted and solid lines delineate adjacent chromatograms. Column, E-1000 micro-Bondagel, mobile phase, 0.1M NaCl; detector, refractive index.] (d) Overlayed chromatograms of dextran standards T-2000, T-500, T-250, T-110, T-70, T-40, T-20, and T-10 [Dotted and solid lines delineate adjacent chromatograms. Column, linear micro-Bondagel; mobile phase, 0.1M NaCl; detector, refractive index.]

that, for either the E-1000 or E-linear column, in 0.05 or 0.1M NaCl, a combination of narrow pullulan standards and broad dextran standards gives universal calibration when $\overline{R}_{\rm gz}$ values are assigned as described in the Experimental section. By way of comparison, as shown by Figs. 4a and 4b, $\log \overline{M}_{\rm w}$ against $K_{\rm av}$ plots, on both columns, and in both mobile phases, gave separate calibration curves for each polysaccharide. In the case of the pullulans, it was assumed that the $K_{\rm av}$ at peak maximum correlated with $\overline{M}_{\rm w}$, whereas, in the case of the dextrans, $K_{\rm av}$ at $\overline{M}_{\rm w}$ was obtained by matching areas as described in the Experimental section.

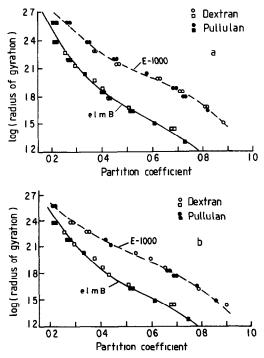


Fig. 3 (a) Universal calibration on columns of E-linear and E-1000 micro-Bondagel in 0 05m NaCl. [All standards in Table I, except T-2000 dextran.] (b) Universal calibration on columns of E-linear and E-1000 micro-Bondagel in 0 1m NaCl [All standards in Table I, except T-2000]

To check on the validity of the universal calibration curves of Figs. 3a and 3b, $\bar{R}_{\rm gz}$ was calculated from the chromatograms of each standard according to Eq. 11.

$$\bar{R}_{gz} = \sum_{i} C_{i} R_{gi}^{2} / \sum_{i} C_{i} R_{gi}$$
 (11)

where C_i is the concentration of the *i*th species.

For the pullulans, as indicated by the data of Table III, the column of E-linear micro-Bondagel (e.l.m.b.) gave better agreement with the literature values for the samples near the total volume, $V_{\rm t}$, whereas the E-1000 columns appeared to give better agreement with literature values for samples near the void value $V_{\rm 0}$. Salt concentration in the mobile phase appeared to have no effect on $\bar{R}_{\rm gz}$; thus, values for the two salt concentrations (i.e., 0.05 or 0.1m NaCl) were combined. In the case of the dextrans, comparison of literature values with measured values gave no appreciable differences with column pore-size distribution or salt concentration in the mobile phase. Therefore, for each dextran sample in Table III, the twelve values of $\bar{R}_{\rm gz}$ were averaged. For polysaccharides that are eluted near $V_{\rm t}$, e.l m.b. columns appear to discriminate better between peaks than E-1000 columns (see Figs. 2a–d), whereas, for those polysaccharides that are eluted near $V_{\rm 0}$, E-1000

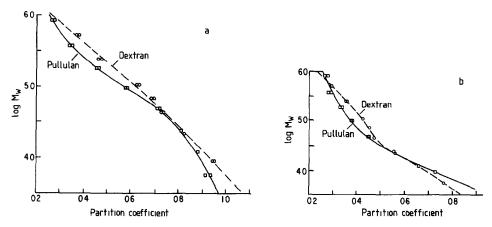


Fig 4 (a) Weight-average molecular weight calibration on column of E-1000 micro-Bondagel in 0 05M NaCl [All standards in Table I, except T-2000 dextran.] (b) Weight-average molecular weight calibration on column of E-1000 micro-Bondagel in 0 1M NaCl. [All standards in Table I, except T-2000 dextran.]

columns appear to discriminate better between peaks than do e.l.m.b., columns. The $\overline{R}_{\rm gz}$ values of the pullulans appear to be more affected by poor resolution at both ends of the size range than are those of the dextrans.

The values of \overline{R}_{gz} , for pullulan, in Table III, were transformed into \overline{M}_w values by utilizing Eq. 8, whereas R_{gz} values for dextrans in Table III were transformed into \overline{M}_w values by utilizing Eqs. 9 and 10. Pullulan and dextran \overline{M}_w values so calculated are given in Table IV. Values of \overline{M}_w from direct molecular-weight calibration (e.g., from calibration curves in Figs. 4a and 4b) were compared with \overline{M}_w values from R_{gz} , even though direct calibration curves contain half of the calibration standards contained in the universal calibration curves. These comparisons revealed that \overline{M}_w from direct calibration (\overline{M}_w cal.) were closer to values provided by the supplier than were values from universal calibration (univ. cal.). Experimental errors in Eq. 8–10 probably account for the additional error in \overline{M}_w from universal calibration, as compared to \overline{M}_w from direct calibration. The \overline{M}_w of T2000 in Table I is only an approximate value. Our data indicate that, for T2000, \overline{M}_w is considerably lower than the nominal value of 2×10^6 .

For the pullulans, it was assumed that the molecular weight at peak maximum did not differ appreciably from $\overline{M}_{\rm w}$. The validity of this assumption was checked by comparing molecular weights at peak maximum with those from the definition of $\overline{M}_{\rm w}$. The data in Table IV confirmed that the assumption is a good one within experimental error (cf., columns 11 and 15).

For the dextrans, \overline{M}_w was correlated with the appropriate K_{av} value by matching areas under the integral weight-distribution curve (e.g., Fig. 1a) with the same integral area obtained from the $\Delta r.i.$ trace the chromatogram (e.g., Fig. 1b). This method of matching areas was checked by calculating the number-average molecular weight obtained from calibration curves such as are found in Figs. 4a and 4b.

TABLE III

RADII OF GYRATION (Å)

Pullulans						Dextrans			;
Sample	Lu value	E.l.m b.b	Rel s.d.º	E-1000	Rel. s d c	Sample	Lut value ^d	E-1000-e 1 m b e	Rel s d f
P-5	20 1	21 3	0.5	263	23	T-10	27 6	28 1	2.0
P-10	31.2	31.8	0.5	33.9	2.5	T-20	42.8	42.0	1.9
P-20	46.3	44.9	0.5	46.7	11.3	T-40	60.5	62.8	18
P-50	70.4	70 0	0.4	74 1	1.0	T-70	75.7	82.7	3.7
P-100	109	115	6.0	109	9.4	T-110	95 6	7 16	2.0
P-200	156	180	4.5	167	19	T-250	139	155	36
P-400	241	271	4.0	168	2.1	T-500	191	202	2.5
P-800	389	414	1.9	375	1.7	T-2000	338	268	63

"Calculated from Eq 8. Measured on E-linear micro-Bondagel column (e I m.b.) Percentage relative standard deviation of 6 determinations (1 column × 2 mobile phase × 3 replicates) "Calculated from Eq. 9 and 10 "Value from e I m b and E-1000 columns combined Percentage relative standard deviation of 12 determinations (2 column × 2 mobile phase concentrations × 3 replicates)

TABLE IV

experimental molecular weights (\times 10^{-3})

Dextrans									Pullulans	\$1					
Sample	M _w Cal ^a	Rs db	Ucʻ	Rsdb	™	Rsd	\mathbf{M}_{p}^{d}	Rsd	Sample	$\overline{\mathbf{M}}_{\mathbf{w}} cal^a$	Rsdb	Uce	Rsdb	\mathbf{M}_{p}^{d}	Rsdb
T-10	10 1	3.0	9 4		98	5 8	9.7	4 1	P-5	57	3.7	54	12	57	91
T-20	19 9	3.0	21 0		158	19	158	30 0	P-10	11.5	33	13.7	29	126	2.7
T-40	42.7	4.7	469		56 9	26	316	8 2	P-20	23 3	120	23 3	14	23.5	12 4
T-70	763	3.4	818	37	42 5	3.1	64 2	9.5	P-50	48 9	13	49 9	14	48 2	3.2
T-110	111	36	114		9 29	41	101	66	P-100	99 2	12	105	60	958	3.1
T-250	243	3.7	330		107	18	161	12.4	P-200	215	47	220	2.0	195	61
T-500	391	5.6	638		174	52	297	12.8	P-400	470	6.1	459	5 5	465	10 4
T-2000	069	5 9	1221		215	13	754	13 3	P-800	780	33	847	3.0	807	10 0

P-5 + P-10 standards, which have values for e 1 m b column only (1 column \times 2 mobile phases \times 3 replicates) or 6 determinations. Number-average molecular weight obtained with log \overline{M}_w calibration dM olecular weight at peak maximum d Universal calibration obtained with log R_{gc} calibration, employing "Obtained with $\log \overline{M}_{w}$ calibration "Percentage relative-standard deviation of 12 determinations (2 columns imes 2 mobile phase imes 3 replicates), except for all standards but T-2000 Generally, these values of \overline{M}_n agree more closely with supplier values than do experimental values of \overline{M}_w compared with supplier values of \overline{M}_w . Surprisingly, comparison of the molecular weight at peak maximum (M_p) values supplied by the manufacturer (see Table I) were very different from the values of M_p found by h.p.s.e.c. (see Table IV).

The only exceptions were the T-20 and T-40 dextrans, for which there was good agreement between the two values. Generally, our $M_{\rm p}$ values from h.p.s.e.c. were higher those provided by the supplier. We assume that the polydispersity of the pullulan samples was sufficiently small that $\overline{M}_{\rm w}$ values supplied by the manufacturer could be associated with the peak maximum observed by h.p.s.e.c. This assumption was proved valid by the data given in Table IV. Values of $\overline{M}_{\rm w}$ from direct molecular-weight calibration agree within experimental error with values obtained from the peak maximum.

CONCLUSION

In conclusion, we have shown that \overline{M}_w correlates well with \overline{R}_{gz} for universal calibration by combining broad and narrow standards in the same calibration curve. Furthermore, well characterized, commercial samples are available for universal calibration of polysaccharides. In addition, direct molecular-weight calibration is somewhat more accurate than are molecular weights from universal calibration. Finally, \overline{R}_g is measured more accurately than \overline{M}_w by universal calibration

REFERENCES

- 1 M L FISHMAN, L A PEPPER, AND P E. PFEFFER, Water Soluble Polymers, Adv Chem Ser, 213 (1986) Ch 3
- 2 M L FISHMAN AND L A. PEPPER, in I C M DEA, V. CRESCENZI, AND T STIVALA, (Eds.), New Developments in Industrial Polysaccharides, Gordon and Breach, New York, 1985, pp 159-166
- 3 M L FISHMAN, L PEPPER, W C DAMERT, J G PHILLIPS, AND R A BARFORD, IN M L FISHMAN AND J J JEN (Eds.), Chemistry and Function of Pectins, A C S. Symp. Ser., 310 (1986) 22-37
- 4 T KATO, T TOKRYA, AND A J TAKOHOSKI, J Chromatogr., 256 (1983) 61
- 5 M L FISHMAN, P E PFEFFER, R A BARFORD, AND L W. DONER, J Agric Food Chem, 32 (1984) 372
- 6 M L FISHMAN, Anal Biochem., 71 (1976) 41
- 7 P F ONYON, in P. W ALLEN (Ed), Techniques of Polymer Characterization, Butterworths, London, 1959, p. 177
- 8 M L FISHMAN, L PEPPER, AND R. A BARFORD, J Polym Sci., Polym Phys Ed., 22 (1984) 899
- 9 W W YAU, J. J. KIRKLAND, AND D BLY, Modern Size-Exclusion Chromatography, Wiley, New York, 1979, p. 293.